Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 293 K Mean σ (C–C) = 0.002 Å R factor = 0.031 wR factor = 0.085 Data-to-parameter ratio = 9.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

(S)-(–)-2,3-Methylenedioxy-8-oxoberbine

In the title optically active protoberberine derivative, $C_{18}H_{15}NO_3$, the absolute configuration has been confirmed as 13aS. The H atom at the asymmetric centre assumes an α orientation and occupies a bisectional position relative to the tetrahydropyridine ring, and is axial to the tetrahydropyridinone ring.

Received 19 October 2005 Accepted 31 October 2005 Online 5 November 2005

Comment

As a continuation of our study of stereoselective syntheses of isoquinoline alkaloids based on the addition of carbon nucleophiles to imines (Brózda et al., 2001), we extended this approach to the synthesis of the protoberberine system. The key step of the synthesis, in which the new stereogenic centre at C13a was created, involved the addition of laterally metallated chiral o-toluamides to cyclic imines. We were interested in performing the first asymmetric synthesis of gusanlung D, using this metallation methodology. (-)-Gusanlung D isolated from Acangelisia gusanlung H. S. Lo by Zhang et al. (1995) is the first optically active natural protoberberine alkaloid unoxygenated at ring D, to which the structure of (S)-(-)-2,3-methylenedioxy-8-oxoberbine was assigned on the basis of spectroscopic analysis. We have obtained (S)-(-)-2,3-methylenedioxy-8-oxoberbine, (I), and its (R)-enantiomer, each with enantiomeric excess >99%, in the reactions of 6,7-methylenedioxy-3,4-dihydroisoquinoline with laterally lithiated o-toluamides in which the amine group was derived from (1S,2S)-thiomicamine or (1R,2S)-2-amino-1phenylpropanol, respectively (Chrzanowska et al., 2004). However our synthetic compound (I) differed significantly in its physical data, viz. specific rotation, melting point and spectroscopic data, from those reported for gusanlung D (Zhang et al., 1995), but are in line with those reported for racemic 2,3-methylenedioxy-8-oxoberbine (Kessar et al., 1992; Reimann et al., 2003). In order to clarify this ambiguity, we have undertaken an X-ray crystallographic study to confirm the correctness of the absolute configuration of synthesized (S)-(-)-2,3-methylenedioxy-8-oxoberbine.



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Figure 1

The molecular structure of (I), showing the atomic labelling scheme. Displacement ellipsoids are drawn at the 30% probability level.

The results of the X-ray study have confirmed the absolute configuration of (I) as S (Fig. 1), as proposed earlier on the basis of spectroscopic data (Chrzanowska et al., 2004). The H atom at the asymmetric centre C13a assumes an α configuration and a bisectional position relative to ring B, and is axial to ring C. The torsion angles H13a-C13a-N7-C6 and H13a-C13a-N7-C8 are 90.1 (10) and -74.6 (10)°, respectively.

Compound (I) is a δ -lactam. As expected, the C8–N7 bond distance of 1.353 (2) Å is comparable with the normal length of the Csp^2 -N bond of 1.352 (3) Å (Allen *et al.*, 1987) for the lactam group $C^*-N(-C^*)-C=0$, where $C^* = Csp^3$. The five-membered heterocyclic ring attached to the sixmembered ring A has the envelope conformation [puckering parameters (Cremer & Pople, 1975) Q = 0.135 (2) Å and $\varphi =$ $325.6 (8)^{\circ}$, with atom C14 deviating from the planar system defined by the other four atoms by 0.215 (3) Å. The sixmembered ring B assumes an envelope conformation [Q =0.493 (2) Å, $\theta = 122.7$ (2)° and $\varphi = 241.5$ (2)°], while ring C has a skew boat conformation distorted towards an envelope [O =0.453 (2) Å, $\theta = 60.9$ (3)° and $\varphi = 275.7$ (2)°]. The difference in the puckering of rings B and C is thought to be mainly a consequence of the presence of the carbonyl group in the latter. The Cambridge Structural Database (CSD, version 5.26; Allen, 2002) contains only one protoberberine alkaloid of a δ lactam type unsubstituted at rings B and C (Aree et al., 2003). In that compound, ring B assumes a conformation intermediate between an envelope and a skew boat, while ring Chas the conformation of a distorted skew boat. Furthermore, the dihedral angle between the least-squares planes through rings B and C is 27.93 (6)°. The dihedral angle made by the planes of the benzene rings A and D is 25.16 $(7)^{\circ}$.

The molecules in the crystal structure of (I) are linked via non-classical C-H···O hydrogen bonds into chains aligned alternately along the different diagonals in the ab plane (Table 1, Fig. 2).





The hydrogen bonding in the crystal structure of (I). The molecular layers at the front and back are drawn with filled and open bonds, respectively. Dashed lines indicate hydrogen bonds.

Experimental

Compound (I) was synthesized according to the literature procedure of Chrzanowska *et al.* (2004); mp. 468–470 K; $[\alpha]_{\rm D} = -432.6^{\circ}$ (*c* 0.80, CHCl₃). Single crystals were grown by slow evaporation of a methanol solution of (I) kept at room temperature. The enantiomeric excess of (I) was established to be >99% by high-performance liquid chromatography analysis using a Chiracel OD-H column.

Crystal data	
Cr ₁₈ H ₁₅ NO ₃ $M_r = 293.31$ Orthorhombic, $P2_12_12_1$ a = 7.3551 (19) Å b = 9.2458 (12) Å c = 20.332 (3) Å V = 1382.7 (4) Å ³ Z = 4	Cu K α radiation Cell parameters from 56 reflections $\theta = 10.1-26.4^{\circ}$ $\mu = 0.79 \text{ mm}^{-1}$ T = 293 (2) K Lath, colourless $0.55 \times 0.17 \times 0.08 \text{ mm}$
$D_x = 1.409 \text{ Mg m}$	

Data collection

Kuma KM-4 diffractometer $\omega/2\theta$ scans Absorption correction: none 2750 measured reflections 2554 independent reflections 2284 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.028$

Refinement

Refinement on F^2
$$\begin{split} R[F^2 > 2\sigma(F^2)] &= 0.031 \\ wR(F^2) &= 0.085 \end{split}$$
S = 1.072554 reflections 260 parameters All H-atom parameters refined $w = 1/[\sigma^2(F_0^2) + (0.0547P)^2]$ + 0.0905P] where $P = (F_0^2 + 2F_c^2)/3$

 $\theta_{\rm max} = 70.1^{\circ}$ $h = -8 \rightarrow 8$ $k = 0 \rightarrow 11$ $l = 0 \rightarrow 24$ 2 standard reflections every 100 reflections intensity decay: 1.4%

 $(\Delta/\sigma)_{\rm max} < 0.001$ $\Delta \rho_{\rm max} = 0.16 \text{ e } \text{\AA}^{-3}$ $\Delta \rho_{\rm min} = -0.11 \text{ e } \text{\AA}^{-3}$ Extinction correction: SHELXL97 (Sheldrick, 1997) Extinction coefficient: 0.0042 (5) Absolute structure: Flack (1983), with 1025 Friedel pairs Flack parameter: 0.0 (2)

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Table 1Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots \mathbf{A}$
$\begin{array}{c} \hline C6-H61\cdots O3 \\ C14-H142\cdots O3^i \end{array}$	1.02 (2)	2.27 (2)	2.722 (2)	105 (1)
	0.99 (3)	2.35 (3)	3.323 (3)	168 (2)

Symmetry code: (i) x - 1, y + 1, z.

All H atoms were located in difference Fourier maps and refined freely; the range of C-H distances was 0.94 (2)-1.03 (2) Å.

Data collection: *KM-4 Software* (Kuma, 1996); cell refinement: *KM-4 Software*; data reduction: *KM-4 Software*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

This work was supported by a research grant from the State Committee for Scientific Research in the years 2003–2006 (KBN grant No. 4 T09A 078 24).

References

- Allen, F. H. (2002). Acta Cryst. B58, 380-388.
- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1–S19.
- Aree, T., Singhana, B., Pakawatchai, Ch., Chavasiri, W. & Kokpol, U. (2003). Acta Cryst. E59, 0919–0921.
- Brózda, D., Chrzanowska, M., Głuszyńska, A., Rozwadowska, M. D. & Sulima, A. (2001). Ann. Polish Chem. Soc. pp. 264–268.
- Chrzanowska, M., Dreas, A. & Rozwadowska, M. D. (2004). Tetrahedron Asymm. 15, 1113-1120.
- Cremer, D. & Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354-1358.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
- Flack, H. D. (1983). Acta Cryst. A39, 876-881.
- Kessar, S. V., Singh, P., Vohra, R., Kaur, N. P. & Venugopal, D. (1992). J. Org. Chem. 57, 6716–6720.
- Kuma (1996). KM-4 Software. Version 8.0.1. Kuma Diffraction, Wrocław, Poland.
- Reimann, E., Grasberger, F. & Polborn, K. (2003). Monatsh. Chem. 134, 991– 1014.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Zhang, J.-S., Le Men-Olivier, L. & Massiot, G. (1995). *Phytochemistry*, **39**, 439–442.